



GAS CONDENSATE OF THE SHURTAN FIELD IS A RAW MATERIAL FOR PETROCHEMICALS

HAMIDOV B.N.¹

¹DR. techn. sciences, Institute of general and inorganic chemistry, academy of sciences of the republic of Uzbekistan, Tashkent

SAMUKOV T.I.²

²cand. techn. sciences, Institute of general and inorganic chemistry, academy of sciences of the republic of Uzbekistan, Tashkent

E-MAIL: TSAMUKOV@GMAIL.COM

<https://doi.org/10.5281/zenodo.7461062>

Annotation

The properties and composition of raw materials - gas condensate of the Shurtan field for the petrochemical industry with a potential content of aromatic, aliphatic and naphthenic hydrocarbons, as well as for the production of solvents of paints and varnishes - were studied. A method of rectification was developed and the mode of processing of gas condensate (HA) was determined and the dissolving abilities for butanol and aniline points of solvents obtained from HA were studied. physicochemical parameters of solvents from HA. Three types of hydrocarbon fraction were used for the analysis: 1-fraction 120-160; 2-fraction 160-220; 3-fraction 120-220.

Key words: Gas condensate, solvent, paint and varnish material, rectification, phlegm, fraction, hydrocarbons, aromatic hydrocarbons.

A potential area of use of solvents is the production and use of paint and varnish materials, which is determined by their successful combination in the dispersions of varnishes and paints, high dissolving ability of various binders (resins), relatively low toxicity, lack of corrosive activity, etc. Solvents are actively involved in the formation of an effective structure and properties of the protective coating (film) [1-2].

To obtain solvents of paints and varnishes (paint and varnish materials), stable GC (gas condensates) of the Shurtan natural gas field were used. Shurtan GC has the following characteristics: 1. A₁₋₂ H₂ F₂₋₃ (1-class; A₁₋₂ - the content of AU in the gasoline fraction; H₂ - the content of alkanes in the diesel fraction; F_{2-z} - fractional composition). The group composition of the Shurtan GC is a peculiar content of AU 25-29% by weight, aliphatic - 45% by weight, naphthenic - 24-25% by weight. The average molecular weight is 164, the refractive index is 1.4417, the specific gravity is 0.769 g/cm³ [3].

Research methods.

System solvents from HA were obtained on the oil rectification apparatus (ARN-2). The selection of the HA fraction was made up to 220 ° C, the following were obtained: gasoline fraction - with 35-150 ° C; solvent of paints of the oil solvent type (Nefras-A-130/150) [4]. with an interval of boiling points from 120-160 ° C, conventionally called GKLM-S3-120/160; heavy oil solvent (Nefras-A-120/200) with an interval of boiling points of hydrocarbons of the GC from 160-220 ° C, conventionally called GKLM-A-160/220 and a fraction of hydrocarbons with a boiling point of 120-220 ° C was obtained, which is a system solvent of the white spirit type (Nefras-C4-155/200), conventionally called GCCM-C4-120/220 [5].

The dissolving abilities of solvents of the GK-LCM series are determined by the effective phase equilibrium of their solutions with a standard resin, that is, Kauri - butanol point (KB) and aniline point (AT) depending on different concentrations of solvents.

Table 1. Physicochemical parameters of solvents from gas condensate

Indicators	Solvents from HA		
	GKKKM - C3- 120/160	GKKKM-A- 160/220	GKKKM - C4- 120/220
Density, g/s m ³	0,864	0,835	0,780
Temperature at the beginning of the kip. °C,	120	160	120
Temperature con. Kip. °C,	160	220	220
Residue in the flask, %,	1,5	1,5	2,0
Flash point in closed crucible, °C,	28	45	41
Volatility on xylene	1,0 - 1,2	3,5 - 4,0	3,0 - 4,5
Aniline point, °C,	49	50	45
Kauri-butanol point, g	54	34	43
Mass fraction, %, wt: aromatic hydrocarbons and sulfur compounds	2,9	27,7	11,9
	0,0062	0,0145	0,0053
Content of mechanical impurities and water	away.	away.	away.
Copper Plate Test	endure.	endure.	endure.
Color on the iodometric scale, mg/l iodine	transparen t	transparent	transparen t
MPC, mg/ m ³	200	150	300

Three fraction samples were used for the analysis: 1-fraction 120-160; 2-fraction 160-220; 3-fraction 120-220. In the study of each fraction, indicators such as mass fraction, mole fraction and octane number of components were obtained.

Results of the study.

Table 2 shows the mass fraction, molar fractions and octane numbers of components of the sample 1-fraction 120-160 obtained from the gas condensate of the Shurtan field of natural gas.

Table 2. Physical and chemical parameters 1-fraction 120-160

C _n \Groups	Mass, %					
	Paraffins	Isoparaffins	Aromatics	Naphthenes	Olefins	Sum
C ₆	0	0,0213087	0	0,0290381	0	0,050
C ₇	0,0323839	0,058821	0,174011	0,444032	0,745985	1,455
C ₈	0,0147179	7,55482	0,0983555	2,04735	0,672183	10,387
C ₉	1,70615	3,15115	0,177065	1,49873	1,19583	7,729
C ₁₀	0	2,62506	1,28202	0	0,671401	4,578
C ₁₁	0,114645	0,860999	0,694287	0	0	1,670
C ₁₂	0,040335	0,187062	0,4905	0,0717787	0	0,790
C ₁₃	0,0660757	0	0	0	0	0,066

Sum	1,974	14,459	2,916	4,091	3,285	26,726
Mole, %						
C_n \ Groups	Paraffins	Isoparaffins	Aromatics	Naphthenes	Olefins	Sum
C ₆	0	0,054842	0	0,0711985	0	0,126
C ₇	0,0969167	0,176037	0,478416	1,30209	2,16293	4,216
C ₈	0,0502095	25,7729	0,311596	6,86154	2,24491	35,241
C ₉	6,53533	12,0704	0,635617	5,65039	4,50842	29,400
C ₁₀	0	11,1543	5,17285	0	2,81259	19,140
C ₁₁	0,535216	4,01954	3,06324	0	0	7,618
C ₁₂	0,205191	0,951613	2,37742	0,360841	0	3,895
C ₁₃	0,363528	0	0	0	0	0,364
Sum	7,786	54,200	12,039	14,246	11,729	100,000
Octane number						
Component	Research Method			Motor method		
Paraffins	1,561			0,788		
Isoparaffins	8,390			7,414		
Aromatics	2,337			2,073		
Naphthenes	3,699			1,966		
Olefins	2,831			1,902		
Oxygenates	0,000			0,000		
Sum	18,819			14,143		

In addition, the molecular weight of the mixture is -33.484 g/mol, the density of the mixture is -194.894 kg/m³, the saturated vapor pressure is -1.034 kPa, the total number of known components is -84.

Table 3 shows the indicators of the mass fraction, molar fractions and octane numbers of the components of the sample 2-fraction 160-220 obtained from the gas condensate of the Shurtan field of natural gas.

Table 3. Physical and chemical parameters 2-fraction 160-220

Mass, %						
C_n \ Groups	Paraffins	Isoparaffins	Aromatics	Naphthenes	Olefins	Sum
C ₇	0	0,00685892	0	0	0,0052018	0,012
C ₈	0	0,0882948	0,067493	0,0052197	0,0771329	0,238
C ₉	0,166521	0,905877	0,130587	1,15342	0,00502853	2,361
C ₁₀	1,30491	3,81055	5,56706	0	0	10,683
C ₁₁	1,48591	1,22877	18,1423	10,3897	1,174	32,421
C ₁₂	0,258342	1,27774	3,84365	0	0	5,380
C ₁₃	0,0659562	0	0	0	0	0,066
Sum	3,282	7,318	27,751	11,548	1,261	51,161

Mole, %						
C _n \ Groups	Paraffins	Isoparaffins	Aromatics	Naphthenes	Olefins	Sum
C ₇	0	0,00921526	0	0	0,0068479	0,016
C ₈	0	0,135225	0,095992	0,0078534	0,116006	0,355
C ₉	0,286353	1,55776	0,210448	1,95221	0,00851096	4,015
C ₁₀	2,48923	7,26896	9,98112	0	0	19,739
C ₁₁	3,11421	2,57528	35,9812	19,7404	2,4287	63,840
C ₁₂	0,59	2,91809	8,36346	0	0	11,872
C ₁₃	0,162904	0	0	0	0	0,163
Sum	6,643	14,465	54,632	21,700	2,560	100,000
Octane number						
Component	Research Method			Motor method		
Paraffins	2,937			2,619		
Isoparaffins	6,351			5,607		
Aromatics	25,007			22,830		
Naphthenes	10,313			9,167		
Olefins	1,138			1,010		
Oxygenates	0,000			0,000		
Sum	45,746			41,232		

Along with the above, the molecular weight of the mixture is -74.586 g / mol, the density of the mixture is -429.275 kg / m³, the saturated vapor pressure is -0.262 kPa, the total number of known components is -87.

Table 4 shows the indicators of mass fraction, molar fractions and octane numbers of components of the sample 3-fraction 120-220 obtained from the gas condensate of the Shurtan natural gas field.

Table 4. Physical and chemical parameters 3-fraction 120-220

Mass, %						
C _n \ Groups	Paraffins	Isoparaffins	Aromatics	Naphthenes	Olefins	Sum
C ₆	0	0,0399284	0	0,0323979	0	0,072
C ₇	0,0363909	0,0508683	0,00700836	0,408687	0,721257	1,224
C ₈	0	1,00669	0,267148	0,82189	1,10558	3,201
C ₉	0,474768	5,04869	1,05997	0,941049	0,0446563	7,569
C ₁₀	0	5,58207	3,18547	0	0,098078	8,866
C ₁₁	0	0,558294	6,23201	0	0	6,790
C ₁₂	0,084716	0,406543	1,15552	0,24716	0	1,894
C ₁₃	0,0313131	0	0	0	0	0,031
Sum	0,627	12,693	11,907	2,451	1,970	29,648

Mole, %						
C _n \ Groups	Paraffins	Isoparaffins	Aromatics	Naphthenes	Olefins	Sum
C ₆	0	0,0858139	0	0,0663344	0	0,152
C ₇	0,0909458	0,127127	0,0160904	1,00078	1,74849	2,983
C ₈	0	2,86785	0,706748	2,30019	3,09312	8,968
C ₉	1,51863	16,1491	3,17743	2,9627	0,140591	23,948
C ₁₀	0	19,807	10,6628	0	0,343096	30,813
C ₁₁	0	2,17649	23,0134	0	0	25,190
C ₁₂	0,359883	1,72704	4,67694	1,03757	0	7,801
C ₁₃	0,143861	0	0	0	0	0,144
Sum	2,113	42,940	42,253	7,368	5,325	100,000
Octane number						
Component	Research Method		Motor method			
Paraffins	0,484		0,272			
Isoparaffins	9,933		7,991			
Aromatics	10,825		9,818			
Naphthenes	2,105		1,440			
Olefins	1,787		1,170			
Oxygenates	0,000		0,000			
Sum	25,133		20,690			

With all this, the molecular weight of the mixture is -40.098 g / mol, the density of the mixture is -235.376 kg / m³, the saturated vapor pressure is -0.618 kPa, the total number of known components is -95.

From the above studies, it can be concluded that, according to the new raw material facility, Shurtan GC and its hydrocarbon fractions in qualitative and quantitative composition are potential and acceptable for the development of technologies for obtaining solvents for paints.

References:

1. Alimov A.A., Samukov T.I., Khodjaeva M.A. Composition of special types of solvents. 1 Republic scientific and technical. Conf. composite materials and their application. Tez.dokl., Tashkent, 1994, - p.438.
2. Drinberg S.A., Itsko E.F. Solvents for paints and varnishes. Reference manual. - Leningrad: Chemistry, Len. otd., 1980, -160 p.
3. Khamidov B.N., Samukov T.I. Development of the selection of components of composite solvents. 1 Republic scientific and technical. Conf. Complex birikmalar kimyoshi wa analyst kimyo fanlarining dolzarb muammolari. Tez.dokl., Termiz, 2022, - p.518.
4. Stekolshchikov M.N. Aliphatic solvent "Nefras" C4-140/200 is an equivalent substitute for White Spirit. Paint and varnish materials and their application. 1987. No 6, - p.52.
5. B.N. Khamidov., Samukov T.I., Preparation of mixed solvents from gas condensate.// Oil refining and petrochemistry.2022.No.6.,-p.28-30.

6. Umbarov, I., Turaev, K., & Samadiy, M. (2020). Research chemical composition of samples of underground salt waters of Surkhandarya and Urtabulok of Bukhara-Karshi depression. *Journal of Kritikal Reviews*, 7(19), 8559-8562.
7. Умбаров, И. А. (2016, November). Комплексная переработка природных подземных вод. In Горно-металлургический комплекс: достижения, проблемы и перспективы инновационного развития: Тез. докл. Республ. науч.-техн. конф.–Навои (р. 366).
8. Умбаров, И. А. (2002). Исследование и усовершенствование технологии получения йода из йодсодержащих подземных вод. Ташкент-2002.-24 с.
9. Умбаров, И. А., & Тураев, Х. Х. (2015). Исследование процесса осаждения йода из раствора абсорбента с потенциометрическим титрованием. *Вестник ТашГТУ*, (4), 151-156.
10. Умбаров, И. А., & Тураев, Х. Х. (2015). Исследование кинетики окисления йодид-ионов в гидротермальных водах различными окислителями. *Узбекский химический журнал*, (6), 12-16.
11. Umbarov, I. A., Mamatraimov, A., & Himmатов, S. (2014). Study of Iodine Distribution in Natural Sources. In *Actual Problems in Analytical Chemistry, Abstract, Proceedings of 4th Republican Research-to-Practice Conference, Termez* (pp. 345-346).
12. Умбаров, И. А., Кулматов, Р. А., Тураев, Н. Й., & Ишанходжаев, С. (2000). Содержание и форма нахождения йода в подземных соленых водах Сурхандарьинской области. *Узб. хим. журн*, (1), 70-72.
13. Ishanhojaev, S., Umbarov, I., Kulmatov, R. A., & Mingturaev, M. (2000). Study of iodine of ions of oxidation by potentiometric method. *UZBEKSKII KHIMICHESKII ZHURNAL*, (6), 14-15.
14. Umbarov, I. Study of iodine of ions of oxidation by potentiometric method. vol. 5 (2021). 2021-06.01.
15. Эшкараев, С. Ч., Тураев, Х. Х., & Умбаров, И. А. (2020). Радиометрическое определение активности бета-излучений стронция-90 в почвах Сурхандарьинской области Узбекистана. In *Вестник научных конференций* (No. 6-1, pp. 121-124). ООО Консалтинговая компания Юком.
16. Эшмуродов, Х. Э., Гелдиев, Ю. А., Тураев, Х. Х., Умбаров, И. А., Джалилов, А. Т., & Бабамуратов, Б. Э. (2020). Получение и исследование модифицированных глифталевых смол с кремнийорганическим соединением. *Universum: технические науки*, (12-5 (81)), 4-8.
17. Умбаров, И. А., Тураев, Х. Х., Набиев, Д. А., Тураханов, М. И., & Холтураев, К. Б. (2019). Процессы выделений йода из концентратов. *Universum: технические науки*, (10-2 (67)), 48-51.
18. Умбаров, И. А., Тураев, Х. Х., Аликулов, Р. В., Чориев, О. Э., & Эшмуродов, Х. Э. (2018). Кинетика окисления йодид-ионов в присутствии различных окислителей из подземных соленых вод. *Universum: технические науки*, (3 (48)), 41-44.
19. Умбаров, И. А., Тураев, Х. Х., Касимов, Ш. А., & Умбарова, М. И. (2017). Усовершенствование способа выделения элементного йода из абсорбентов. *ILMIY AXBOROTNOMA*, 5.
20. Умбаров, И. А., Тураев, Х. Х., Касимов, Ш. А., & Умбарова, М. И. УСОВЕРШЕНСТВОВАНИЕ СПОСОБА ВЫДЕЛЕНИЙ ЭЛЕМЕНТНОГО ЙОДА ИЗ КОНЦЕНТРАТОВ.

21. Курбанов, Ш. Й., Умбаров, И. А., & Бабамуратов, Б. Э. (2022). ОБЛАСТИ ПРИМЕНЕНИЯ ЙОДА И ЙОДНЫХ СОЕДИНЕНИЙ. In НАУКА И МОЛОДЁЖЬ: АКТУАЛЬНЫЕ ВОПРОСЫ СОВРЕМЕННЫХ НАУЧНЫХ ИССЛЕДОВАНИЙ (pp. 12-15).
22. Умбаров, И. А. (2021). Количественное Определение Содержания Йода В Гидротермальных Подземных Водах. CENTRAL ASIAN JOURNAL OF MEDICAL AND NATURAL SCIENCES, 2(6), 339-342.
23. Гелдиев, Ю. А., Тураев, Х. Х., Умбаров, И. А., Эшмуродов, Х. Э., & Джалилов, А. Т. (2021). СИНТЕЗ И ИССЛЕДОВАНИЕ НОВОГО СОЕДИНЕНИЯ НА ОСНОВЕ ПОЛИКРЕМНИЕВОЙ КИСЛОТЫ, МОДИФИЦИРОВАННОЙ МОНОЭТАНОЛАМИНОМ. Universum: химия и биология, (10-2 (88)), 78-81.
24. Эшанкулов, Х. Н., Тураев, Х. Х., Умбаров, И. А., & Джалилов, А. Т. (2021). ОРГАНИЧЕСКИЙ ПОЛИМЕРНЫЙ СИНТЕЗ МЕТАЛЛОВ НА ОСНОВЕ АКРИЛОНИТРИЛА. Universum: технические науки, (7-3 (88)), 5-8.
25. Гелдиев, Ю. А., Тураев, Х. Х., Умбаров, И. А., & Джалилов, А. Т. (2021). СИНТЕЗ И ИК-СПЕКТРОСКОПИЧЕСКИЙ АНАЛИЗ ПРОИЗВОДНЫХ ПОЛИКРЕМНИЕВОЙ КИСЛОТЫ С МОЧЕВИНОЙ И ФОРМАЛЬДЕГИДОМ. Universum: химия и биология, (7-1), 95-98.
26. Umbarov, I. A., & Turaev, H. K. INTERNATIONAL JOURNAL OF ENGINEERING SCIENCES & RESEARCH TECHNOLOGY INVESTIGATION MECHANISMS AND KINETICS OF OXIDATION IODIDE IONS IN HYDROTHERMAL WATERS VARIOUS OXIDANTS.
27. Umbarov, I. A., & Turaev, H. K. INTERNATIONAL JOURNAL OF ENGINEERING SCIENCES & RESEARCH TECHNOLOGY STUDY OF IODINE DISTRIBUTION FORM IN AN ABSORBENT.
28. Умбаров, И. А., Тураев, Х. Х., Касимов, Ш. А., & Умбарова, М. И. (2017). Усовершенствование способа выделения элементного йода из абсорбентов. ILMII AXBOROTNOMA, 5.
29. Potaenko, K. D., Umbarov, J., Kulmatov, R. A., Turayev, N. Y., & Ishankhodjayev, S. (1999). The removal of iodine from underground salt water by the air desorption method. UZBEKSKII KHIMICHESKII ZHURNAL, (5/6), 92-94.
30. Kulmatov, R. A., Turaev, N. Y., Kenjaev, D., Tillaev, K., & Umbarov, I. (1999). Investigation of distribution and migration forms of toxic elements by complex radioactivation methods.
31. Kulmatov, R. A., Kenjaev, D. S., Umbarov, I., Tillaev, K., & Normuradov, B. (2001). The investigation of physic-chemical forms of toxic metals by activation analysis.
32. Умбаров, И. А., & Тураев, Х. Х. (2017). Потенциометрические исследования окисления ионов йода с нитрита натрия. Universum: технические науки, (12 (45)), 48-50.
33. Умбаров, И. А., & Тураев, Х. Х. (2017). КОМПЛЕКСНАЯ ПЕРЕРАБОТКА ГИДРОТЕРМАЛЬНЫХ ВОД. In EUROPEAN RESEARCH: INNOVATION IN SCIENCE, EDUCATION AND TECHNOLOGY (pp. 8-11).
34. Умбаров, И. А., & Тураев, Х. Х. (2018). ОПРЕДЕЛЕНИЕ ЭЛЕМЕНТНОГО СОСТАВА ПОДЗЕМНЫХ СОЛЕННЫХ ГИДРОТЕРМАЛЬНЫХ ВОД. Science Time, (2 (50)), 76-79.
35. Умбаров, И. А., & Тураев, Х. Х. (2018). Подземные воды как промышленное ценное сырье. Universum: технические науки, (1 (46)), 7-9.
- Умбаров, И. А., & Тураев, Х. Х. (2019). Изучение элементного состава гидротермальных подземных соленых вод скважины " Каттакум-2" месторождения Хаудаг. Universum: химия и биология, (9 (63)), 9-12.